Physicochemical, Volatile Organic Composition, Phenolic, Flavonoid and Ascorbic Acid Contents, Antioxidant, Anti-Arthritic and Anti-Inflammatory Properties of *Cocos nucifera* Juice

By Ololade, Z.S., Kuyooro, S.E., Ogunmola, O.O. & Oyelese, O.J.

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**Abstract**- Various parts of *Cocos nucifera* are locally used for treatment of diseases and production of some foods and beverages for man and animals. This study examined the physicochemical properties, phytochemical and multi therapeutic potentials of juice of *C. nucifera* from Nigeria. These were measured using GC-MS, pH meter, specific gravity, UV-Vis spectrometry, Folin-Ciocalteu’s, aluminium chloride, DPPH, PTAC and egg albumin methods respectively. GC-MS analysis revealed the presence of nitroisobutylglycerol as the most abundant volatile organic compounds in the juice. The pH, clarity, turbidity, TPC, TFC and TAA were 5.09, 1.34, 1.07, 2,261.5±0.00 μgmg⁻¹ GAE, 20.00±0.0 μgmg⁻¹ QE and 66.75±0.00 μgmg⁻¹ AAE, respectively. The antioxidant IC₅₀ and AAI values of the juice were 0.25 mgml⁻¹ and 160 and it was capable of scavenging free radicals at a range between 33.09-76.26%. The TAC was 645.38±0.00 μgmg⁻¹ AAE. The protein denaturation inhibition capacity was at the range between 42.0-73.4%. Therefore, the juice of *C. nucifera* could be used as multi therapeutic agent.

**GJMR-B Classification**: NLMC Code: QV 55
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I. Introduction

Natural products are good drug leads and phytochemical probes to explore mechanisms associated with infection of diseases (Kingston, 2011; Ibekwe and Ameh, 2014). Numerous plant polyphenols reportedly afford multiple health benefits. They are regarded as healthy food and nutrient sources because of their many beneficial components (Voravuthikunchai and Howe, 2014; Zhang et al., 2016). Secondary metabolites in the natural products are considered to be responsible for positive health outcomes (Gechev et al., 2014). Particularly, it is widely noted that plants produce a great deal of antioxidants to combat the oxidative stress induced by oxygen and light in the natural environment (Cartea et al., 2011; Li et al., 2016). Natural products possess antimicrobial and antioxidant activities responsible for the prevalence of dermatophytosis (Thebo et al., 2016). Many approved therapeutics and drugs are derived from natural sources (Cragg and Newman, 2013; Lahrou, 2013).

Cocos nucifera Linn commonly known as coconut is an important fruit tree in the tropical regions and the fruit can be made into a variety of foods and beverages (Yong et al., 2009). C. nucifera is an important member of the family Arecaceae. The juice of C. nucifera has the ability to prevent diseases and sickness. This is due to the free radical scavenging abilities of the antioxidant phytochemicals in it. C. nucifera has long been used in traditional medicine for different kind of illness and almost all parts have their uses. The juice inside the fruit is sterile but when it is extracted and exposed to air, it becomes subjected to quick oxidation and microbial contamination leading to depletion of nutrients and spoilage (Matsui et al., 2008; Queiroz et al., 2008; Jean et al., 2009; Nakono et al., 2012; Adubofuor et al., 2016). Juice of C. nucifera is one of the natural food products to quench thirst and easily available in most of the countries. Both water and meat of coconut refresh the body by providing nutritious content. Traditionally, it has been used to protect the body against infection by dangerous diseases. It has been found to improve digestion and hasten the absorption of nutrients including vitamins, minerals, and amino acids. Recently, the health and medicinal uses of C. nucifera products get research interest because it contains several metabolites such as sugars, proteins, free amino acids, vitamins, minerals and growth promoting factors (Reddy and Lakshmi, 2014). The juice of C. nucifera contains many enzymes including acid phosphatase, catalase, dehydrogenase, diastase, peroxidase and RNA polymerase. Juice of C. nucifera is locally consumed fresh, directly from the fruit. (Adubofuor et al., 2016). This study aimed at evaluations of physicochemical, phytochemical and therapeutic efficacies of C. nucifera juice.

II. Materials and Methods

The fruit of C. nucifera was gotten from Ota, Ogun State, Nigeria and the juice was collected and then stored in vial at 5 °C temperature to prevent contamination.
a) Determination of Clarity and Turbidity
Clarity and turbidity of the juice was determined by measuring the absorbance at 525 and 660 nm respectively using a UV-Vis spectrophotometer (Surajbhan et al., 2012).

b) Colour Determination
Colour of the juice was determined by physical observation in day light (Barkatullah et al., 2012).

c) Odour Determination
Odour of the juice was determined by organoleptic evaluation (Aloko et al., 2017).

d) Determination of pH
The pH of the juice of C. nucifera was determined immediately after extraction at room temperature using digital pH meter (Paz et al., 2016).

e) Determination of Specific Gravity (SG)
A clean specific gravity bottle was weighted (W₀). Then the bottle was filled to the brim with water and stopper was inserted. The water on the stopper and bottle were carefully wiped off and reweighed (W₁). Same process was repeated, but using juice samples instead of water and weighted again (W₂). The specific gravity of the juice was calculated using the formula below.

Specific gravity = (W₂-W₀)/(W₁-W₀)

Where:
W₀ = Weight of empty specific gravity bottle
W₁ = Weight of water + specific gravity bottle
W₂ = Weight of test sample + specific gravity bottle.

f) GC-MS Analysis
The juice of C. nucifera dissolved in methanol was analysed using Shimadzu GC-MS-QP2010 Plus (Japan). The separations were carried out using a Restek Rtx-5MS fused silica capillary column (5%-diphenyl-95%-dimethylpolysiloxane) of 30 m× 0.25 mm internal diameter (di) and 0.25 mm in film thickness. The conditions for analysis were set as follows; column oven temperature using the NIST computer data base matching their mass spectra with those of the end m/z, 700. Detector was operated in EI ionization mode with total flow rate 45.0 ml/min; interface temperature, 200 °C; column flow rate, 0.99 ml/min; ion source temperature, 250 °C; solvent cut time, 3.0 min; start time 3.5 min; end time, 24.0 min; start m/z, 50 and end m/z, 700. Detector was operated in EI ionization mode with total flow rate 45.0 ml/min; column flow rate, 0.99 ml/min; ion source temperature, 250 °C; solvent cut time, 3.0 min; start time 3.5 min; end time, 24.0 min; start m/z, 50 and end m/z, 700. Detector was operated in EI ionization mode with total flow rate 45.0 ml/min; column flow rate, 0.99 ml/min; ion source temperature, 250 °C; solvent cut time, 3.0 min; start time 3.5 min; end time, 24.0 min; start m/z, 50 and end m/z, 700.

g) Determination of Total Phenolic Content (TPC)
The TPC of the juice of C. nucifera was determined using Folin-Ciocalteau method. 1 ml of juice was mixed with 1 ml of 10% Folin-Ciocalteu reagent in distilled water and then neutralized with 4 ml of 7.5% sodium carbonate solution. The sample was maintained at room temperature for 3 hrs with periodical mixing, the absorbance at 760 nm was measured using UV-Vis spectrophotometer. Gallic acid was used as a reference and for the calibration curve; result was expressed in micrograms per gram of gallic acid equivalent (Vasudevarao and Sravanthi, 2017).

h) Total Flavonoid Concentration (TFC)
The total flavonoid content of the juice of C. nucifera was determined by spectrophotometry, using aluminum chloride method. Briefly, 1.0 ml of the juice, 0.10 ml of 10% aluminium chloride, 0.10 ml of sodium acetate (1 M) and 2.80 ml of distilled water. After incubation for 40 min, absorbance was measured at 415 nm using a UV-Vis-spectrophotometer. To calculate the concentration of flavonoids, we prepared a calibration curve using quercetin as standard. The index of TFC concentration is expressed as quercetin equivalents (QE) in µg per mg of juice. All assays were carried out in triplicate (Formagio et al., 2015).

i) Determination of Free Radical Scavenging and Antioxidant Activities
i. In vitro 2,2ʹ-Diphenyl-1-picryl-hydrazyl Assay
The antioxidant and free radical scavenging of the juice of C. nucifera was measured by using DPPH. Briefly, the reaction mixture of 2.0 ml; consist of 1.0 ml of DPPH in methanol (0.004%) and 1.0 ml of various concentrations of juice. Then incubated for 30 min. in dark, and the absorbance was measured at 517 nm. The control was prepared by DPPH and methanol in place of sample. In this assay, the positive control is ascorbic acid. The percentage of inhibition can be calculated using the formula:

\[
I\% = \frac{[A_{blank} - A_{juce}]}{A_{blank}} \times 100
\]

Where: \(A_{blank}\) is the absorbance of blank solution and \(A_{juce}\) is the absorbance of the juice. The dose response curve was plotted and IC₅₀ value for the juice and the standard were calculated (Ololade et al., 2016).
Antioxidant Activity Index: The antioxidant activity index (AAI) was calculated as:

\[
\text{AAI} = \frac{[\text{DPPH initial concentration}]}{[\text{IC}_{50}]}
\]

AAI was classified as weak, when AAI < 0.5, moderate, when AAI ranged between 0.5-1.0, strong, when AAI ranged between 1.0-2.0, and very strong, when AAI > 2.0 (Arulpriya and Lalitha, 2014).

ii. Phosphomolybdate Total Antioxidant Capacity (PTAC) Assay

The PTAC of the juice of C. nucifera was determined with phosphomolybdenum using ascorbic acid as the standard. An aliquot of 1.0 ml of juice solution is combined with 1.0 ml of reagent (0.6 M sulphuric acid, 28 µM sodium phosphate and 4 µM ammonium molybdate). The tubes were capped and incubated in a boiling water bath at 95 °C for 90 min. after the samples had cooled to room temperature, the absorbance of the aqueous solution of each were measured at 695 nm in UV spectrophotometer. The blank solution contained 1.0 ml of reagent solution and the appropriate volume of the same solvent used for the sample and it was incubated under same conditions as the rest of the samples. The total antioxidant capacity was expressed as equivalents of ascorbic acid (Bulus et al., 2017).

In-vitro Anti-Arthritic and Anti-Inflammatory Activities of the Juice on Inhibition of Protein Denaturation (Egg Albumin Assay): in vitro anti-arthritic/anti-inflammatory activity of the juice was evaluated against protein denaturation method using fresh hen’s egg albumin. About 5 ml reaction mixtures (0.2 ml of egg albumin, 2.8 ml of phosphate buffered saline (PBS, pH 6.4) add 2 ml of test sample at 1000, 500, 250 and 125 µg ml−1). Distilled water with same volume (0.2 ml) was used as control. The mixtures were incubated at 37 °C in BOD incubator for about 15 min. followed by heating at 70 °C for 5 min. After cooling to the room temperature, absorbance was measured spectrophotometrically at 660 nm using vehicle as blank. Aspirin (3000 µg ml−1) was used as reference drug. The drug concentration for 50% inhibition (IC_{50}) was determined by plotting percentage inhibition with respect to control against treatment concentration. (Smitha et al., 2017).

III. RESULTS AND DISCUSSION

a) Physicochemical Properties of the Juice of C. nucifera

Determination of different physicochemical properties showed the practical importance and provides bases for suitability, consumption, utility, nutritional and physical qualities of the natural juice of C. nucifera in daily life (Angaye and Maduelosi, 2015). Physicochemical properties of the juice such as colour, odour, pH, clarity, turbidity, specific gravity (Table 1) showed the quality of the juice of C. nucifera from Nigeria.

b) Colour and Odour of the Juice of C. nucifera

The colour of the fresh juice of C. nucifera was milky in nature with sweet aromatic odour.

c) Clarity and Turbidity of the Juice of C. nucifera

The clarity and turbidity of the juice of C. nucifera were determined as 0.74 and 0.66 respectively using UV-Vis spectrophotometer.

d) pH of the Juice of C. nucifera

The pH value of the C. nucifera juice was 5.09; which was within the standard limit (pH 3.40–6.10) that insures freshness of the juice (El-Soahamy et al., 2015). The acidity of juice might be due of present organic acid in the juice, which is responsible for important characteristics of juice: flavour and stability against microbial spoilage and this may confer longer keeping quality of the juice (Nadzirah et al., 2012; Offia-Olua and Ekwunife, 2015). Furthermore, it might also indicate that the juice of C. nucifera have high content of minerals. pH is a very important parameter in the conduct of fermentation. A pH of 4 is the optimum for the growth of fermentative yeast. This also inhibits the development of undesirable microbial flora (Ahoussi et al., 2015; Walker and Stewart, 2016).

e) Specific Gravity of the Juice of C. nucifera

The specific gravity of the juice of C. nucifera was 1.01 (Table 1). This was in line with the amount stated for beverages (including soft drinks and juices) and fruit drinks (low calories and undiluted) as 1.01-1.03. The more sugar present in a juice, the denser the juice becomes. Juice is low dense foods because of its high water content, which provides high volume and weight. To stay within low density guidelines, it is important to either consume natural fruit juice that has not been dehydrated than to eat processed fruit juice that contains added sugar (Swinburn et al., 2004; Ledikwe et al., 2006; Slavin and Lloyd, 2012; Babajide et al., 2013).
**Table 1:** Colour, Odour, Turbidity, pH, and Specific Gravity of the Juice of *C. nucifera*

<table>
<thead>
<tr>
<th>Colour</th>
<th>Odour</th>
<th>Clarity</th>
<th>Turbidity</th>
<th>pH</th>
<th>SG</th>
</tr>
</thead>
<tbody>
<tr>
<td>Milky</td>
<td>Sweet aromatic smell</td>
<td>0.74</td>
<td>0.66</td>
<td>5.09</td>
<td>1.01</td>
</tr>
</tbody>
</table>

**Table 2:** TPC, TFC and TAA of the Juice of *C. nucifera*

<table>
<thead>
<tr>
<th>TPC (mg mg⁻¹ GAE)</th>
<th>TFC (mg mg⁻¹ QE)</th>
<th>TAA (mg mg⁻¹ AAE)</th>
</tr>
</thead>
<tbody>
<tr>
<td>2.261.5 ± 0.00</td>
<td>20.0 ± 0.00</td>
<td>66.75 ± 0.00</td>
</tr>
</tbody>
</table>

Data are presented as the mean ± S.D. of triplicate.

**Table 3:** Antioxidant Properties of the Juice of *C. nucifera*

<table>
<thead>
<tr>
<th>Juice and Reference Drug</th>
<th>DPPH IC₅₀ (µg mg⁻¹)</th>
<th>AAI</th>
<th>PTAC (µg mg⁻¹ AAE)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Juice</td>
<td>0.25</td>
<td>160</td>
<td>645.38 ± 0.00</td>
</tr>
</tbody>
</table>

**j) Organic Composition**

GC-MS analysis revealed the presence of nitrosobutylglycerol (C₆H₇NO₃) as the most abundant volatile organic compound of the juice. The mass spectrum of the compound with retention time 14.924 and retention index 1444 gave 8 major peaks (m/z) at 27, 29, 31, 55, 57, 73, 85, and 86. Nitrosobutylglycerol is a low molecular weight (151) medicinal compound with oxygen-induced, antioxidant, anti-staphylococcal activities (Rane and Anusha, 2012).

**g) Total Phenolic Content, Total Flavanoid Contents and Total Ascorbic Acid**

The TPC, TFC and TAA analyses of the investigated juice of *C. nucifera* showed the presence of high amount phenolic, flavonoid compounds and ascorbic acid (Table 2). Natural phenolic compounds and ascorbic acid play many significant roles in human health as evident from their therapeutic properties (Dimitros, 2006; Ansari et al., 2013; Dzialo et al., 2016). Plants consumed by humans may contain thousands of different amounts of ascorbic acid, phenolic and flavonoid components (Saxena et al., 2013; Kasote et al., 2015; Zhang et al., 2015). The effect of dietary ascorbic acid and phenolics is currently of great interest due to their antioxidant and possible anticarcinogenic activities (Nahak et al., 2014; Pereira et al., 2009).

Ascorbic acid, Phenolic and flavonoid compounds are chain breaking antioxidant, free radicals scavenger and quenchers of singlet oxygen formation in the process of formation of intracellular substances throughout the body (Mitra and Uddin, 2014; Ozcan et al., 2014). Moreover, ascorbic acid, phenolic and flavonoid components play important roles in the control of cancer and other human diseases (Ghasemzadeh and Ghasemzadeh, 2011).

**h) Free Radical Scavenging and Antioxidant Potentials**

The percentage inhibitions of the juice at various concentrations (1000, 750, 500 and 250 µg ml⁻¹) were 76.26, 71.22, 71.00 and 33.09% respectively; while the IC₅₀ value was found to be 0.25 mg ml⁻¹ in comparison to ascorbic acid with IC₅₀ value of 9.0 µg ml⁻¹.

**i) Phosphomolybdate Total Antioxidant Capacity (PTAC)**

The PTAC of the juice of *C. nucifera* was 645.38 ± 0.00 µg mg⁻¹ AAE (Table 3). The measure of the ability of natural products to delay oxidative stress in a controlled system is defined as total antioxidant capacity (Apak et al., 2016; Zhang et al., 2016; Pieme et al., 2017; Tyagi and Agarwal, 2017). The juice showed high antioxidant potential and this can be related to the high amounts of ascorbic acid, flavonoids and phenolic compounds in juice. Antioxidant play definite roles in many pathological conditions and they are known to fight against these free radicals and protects body from various diseases (Aprioku et al., 2013; Lone et al., 2013). Their mechanism of action is either by scavenging the reactive oxygen species or protecting the antioxidant defence mechanisms (Birben et al., 2012). The total antioxidant potential is a relevant tool for investigating the relationship between dietary antioxidants and pathologies induced by the oxidative stress (Pisochi and Negulescu, 2011).

**j) In-vitro Anti-Arthritic and Anti-Inflammatory Potential**

The juice of *C. nucifera* possesses potentially useful anti-arthritic and anti-inflammatory activities at the doses tested. The juice exhibited significant inhibition of egg albumins denaturation of protein between 42.0-73.4% at concentrations between 125-1000 µg ml⁻¹ and with the IC₅₀ value of 6.0 µg ml⁻¹. The result was comparable to that of standard drug aspirin (89.4%) (Table 4). The juice showed inhibition of heat-induced protein (albumin) denaturation and prominent effects on protein denaturation was produced. Natural products that can prevent protein denaturation would be very useful for the development of anti-arthritic and anti-inflammatory drugs (Sowjanya et al., 2013; Janakiraman and Parmeswari, 2014; Obaseki et al., 2016). Therefore, the juice is a promising anti-arthritic agent of natural origin in the treatment of inflammatory disorders. It shows that the juice is capable of reducing the production of auto-antigen which indirectly reduces the protein denaturation and hence alleviate arthritis (Alamgeer et al., 2017; Boddupally et al., 2017). Protein denaturation is one of the leading causes of inflammatory as well as arthritic diseases, which led to production of auto antigens, progressing to certain rheumatic diseases (Jayaprakasam and Ravi, 2012; Pashikanti et al., 2014; Elisha et al., 2016; Mahabaland...
Kaliwal, 2017). The main mechanism involved in protein denaturation is characterized by changes or alterations in hydrophobic, electrostatic, hydrogen and disulphide bonding among the protein molecules (Zavodszyk et al., 2001; Sangeetha and Vidhya, 2016).

Table 4: Egg Albumin Anti-Arthritic/Anti-Inflammatory Activity of the Juice of C. nucifera and Reference Drug

<table>
<thead>
<tr>
<th>Conc. µgml⁻¹</th>
<th>% Inhibition</th>
<th>IC₅₀ µgml⁻¹</th>
<th>% Inhibition of Aspirin 3000 µgml⁻¹</th>
</tr>
</thead>
<tbody>
<tr>
<td>1000</td>
<td>73.4</td>
<td>6.0</td>
<td>89.4</td>
</tr>
<tr>
<td>500</td>
<td>71.4</td>
<td></td>
<td></td>
</tr>
<tr>
<td>250</td>
<td>48.0</td>
<td></td>
<td></td>
</tr>
<tr>
<td>125</td>
<td>42.0</td>
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IV. Conclusion

The results of this study showed that the juice of C. nucifera can be an accessible source of promising therapeutic agents that can be used in combating some infectious diseases caused. The study showed the presence of significant antioxidant, anti-arthritic anti-inflammatory activities of the juice. The activities were due to the presence of pharmacologically active phytochemicals in the juice. Fresh juice of C. nucifera is a fluid that could be consumed for health, refreshment and nutritional purposes. Therefore the juice can include the production and commercialization of foods and drugs.

Conflict of Interest Statement: The authors declared that the research was conducted in the absence of any commercial or financial relationships that could be construed as a potential conflict of interest.

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Dimethyl Sulfone, *IOSR Journal of Pharmacy and Biological Sciences*, 12(1), 93-104.

