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# Phenolic-Protein Interaction: Effects on Functional Properties of Phenolics and Advantages on Phenolic Delivery Platform Development

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#### 8 Abstract

<sup>9</sup> The interaction between phenolics and proteins is one of the most research areas in food,

<sup>10</sup> pharmaceutical, and cosmetic fields since these interactions affect the functional properties of

<sup>11</sup> proteins and phenolics. By using this phenomena, organoleptic properties of phenolic

<sup>12</sup> compounds can be engineered, foaming and gelling properties of proteins can be increased,

<sup>13</sup> and also novel delivery platforms for phenolics can be designed. During the construction of

14 phenolic-protein interaction, both covalent and non-covalent bonds occur. Formation of these

<sup>15</sup> bonds depends on the type of phenolic compounds and protein and also environmental

 $_{16}$   $\,$  conditions such as pH, temperature and ionic strengths. Understanding the exact mechanism

<sup>17</sup> behind these interactions are leading to generate tunable food, drug, and cosmetic products.

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19 Index terms— interaction, antioxidant capacity, bio accessibility, sensory quality, delivery of phenolics.

#### <sup>20</sup> 1 Introduction

21 henolic compounds, abundantly present in plants, account for one-third of the dietary phenols and are a large family of phytochemicals (Herrmann & Nagel, 1989). The phenolic compounds have numerous physiological 22 23 functions, such as anti-inflammatory, antimutagenic, and antioxidant properties through phenolic compounds' 24 potential to protect from oxidative stress (Lee et al., 2005; Altin, Gültekin-Özgüven & Ozcelik, 2018). The studies on animals and also human submits that phenolic compounds in daily diet have significant roles in protection 25 from several diseases including certain types of cancers, cardiovascular diseases, and prevention of osteoporosis 26 27 (Ali, H., 2012). Even insignificant structural differences in the location, number, or form of substituted groups can directly affect the bio-distribution, free concentration, and metabolism of phenolic compounds, and resulting 28 affect to their bioactivity (Jaldappagari et al., 2013). 29

Interactions between the different food compounds have been commonly identified and it is known to have 30 influence on the biological, functional and, nutritional properties of food products. The interaction of phenolic 31 compounds-proteins seems to be the most important one using different aspects. Phenolic acids could be 32 interacted by non-covalently and covalently with proteins, and the interactions of protein-phenolic may affect 33 34 the biological and functional properties of phenolics as well as proteins (Charlton et al., 2002). The bioactivity 35 of a compound after the consumption is related to its bioaccessibility and bioavailability. Based on the type of 36 interaction between phenolics-proteins can be increased or decreased their bioaccessibility and bioavailability. In the sensory aspect, organoleptic quality of food product such as color, bitterness, and astringency, is directly 37 affected form phenolic-protein interaction. (De Freitas & Mateus, 2012; Chung, C., Rojanasasithara, T., 38 Mutilangi, W., & McClements, D. J., 2017). 39

In last decade, numerous studies are focusing on delivery of phenolics, since to contribute the functionality of phenolics is crucialsince they have several positive effects on health such as antioxidant, antimicrobial, antiinflammatory, anti-carcinogenic, and hepatoprotective effect. Protein-phenolic conjugates and protein-based

#### 2 III. INTERACTION EFFECT ON NUTRITIONAL AND ANTIOXIDANT QUALITY OF FOODS

nanoparticles are the most common studied techniques about a delivery of phenolics, which can be used in food,
pharmaceutical, and cosmetic sector.

Therefore, understand the exact mechanism between phenolic-protein interaction not only leads to the 45 observation of nutritional and sensory changes on foods but also enhance to develop novel strategies for functional 46 foods and dietary supplement area. While the interactions can be analyzed considering several aspects, it still 47 is a challenge for food analysis and researchers (Czubinski, J., & Dwiecki, K., 2017). To comprehend the 48 roles of proteins and phenols in interaction, it is essential to designate the nature of the physicochemical and 49 chemical interactions of the proteins-phenolic acids (Ali, H., 2012). The bonding types can be characterized by 50 spectroscopic methods, microscopic methods, thermodynamic methods, bioinformatics methods, electrophoretic 51 52 and chromategraphic methods.

The goal of this review is to give researchers an overview of the currently used methods for identification of bonding type, negative and positive effects of interaction on food quality, bioaccessibility, bioavailability as well as to introduce the novel delivery strategies that base on phenolic-protein interaction.

56 II. Type of Bindings that Contributes to the Interaction

The phenolic-protein interaction is mainly contributed via non-covalent bonds, which are weaker than covalent 57 58 bonds and they are always reversible. Among the covalent bonds, hydrogen bonds provide more stable complex 59 than Van-der -Waals interactions, dipole-dipole interactions and hydrophobic interactions (Yuksel et al., 2010; 60 ??agy et al., 2012; ??akobek 2015). While non-covalent bonds commonly occur in proteinphenolic interaction, in 61 some case covalent interaction can also be formed (Gallo et al., 2013;El-Maksoud et al., 2018;Sui et al., 2018). To determine the phenolic protein interaction is necessary for the biological activity of phenolics as well as proteins. 62 There are several approaches to investigate this interaction such as, spectroscopic methods, microscopic methods, 63 thermodynamic methods, bioinformatics methods, electrophoretic and chromatographic methods. Several studies 64 focus on to investigate phenolic-protein interaction. Previous studies and their findings are summarized in Table 65 1. To obtain the excat result during determination of the binding type, different techniques are used together 66 (Table 1). 67 The proteomic approach is one of the novel bioinformatic technique to identify the binding type. Gallo et al. 68 (2013), investigated the type of interactions between cocoa polyphenols and milk proteins by proteomic technique. 69 They characterized the interaction of ?-lactoglobulin (?-Lg) with catechin and epicatechin, moreover identified the 70 amino acid residue at the binding site. For this aim, they used the matrix-assisted laser desorption ionization-time 71 72 of flight mass spectrometry (MALDI-TOF-MS) and the electrospray ionization tandem quadrupole/orthogonal-73 acceleration time-of-flight mass spectrometer (ESI-Q-TOF MS/MS). In these analyses, tryptic peptides of ?-Lg have allowed the identification of the binding site as the free thiol group of cysteine and found that while 74 polyphenols covalently bound with ?-Lg, they interact with casein via non-covalent bonds. In another study, 75 fluorescence, circular dichroism spectroscopy, and docking studies were used for characterization of the interaction 76 between ?-Lg and cyanidin-3-O-glucoside According to this study, the interaction was mainly contributed by 77 both hydrogen bonding and the hydrophobic interaction (Cheng et al., 2017). Furthermore, after the binding of 78 cyanidin-3-O-glucoside to ?-Lg, the secondary structure of the ?-Lg was changed in which the major structure 79 of ?-sheetincreased and the minor structure of ?-helix decreased. The changes in secondary structure of proteins 80 after the phenolic interaction also identified in different studies (Zhang et al., 2014;Al-Hanish et al., 2016;Jia et 81 al., 2017). The significant reduction of ?helix and an increase of ?-sheet and turn structures were determined in 82 ?-lactalbumin and ?-Lg in the phenolic-protein complex (Zhang et al., 2014). Also, Jia et al. (2017) observed the 83 conformational changes on the secondary structure of ?-Lg, after phenolic compound attachment, via circular 84 dichroism and Fourier transform infrared. According to their findings, after the binding of the phenolic compound, 85 the surface hydrophobicity of ?-Lg was changed. Thus ?-helix to ?-structures transition occurred. The noncovalent 86 interactions between ?-lactalbumin and epigallocatechin-3-gallate were determined by circular dichroism and 87 Fourier transform infrared spectroscopy (Al-Hanish et al., 2016). Based on their findings, epigallocatechin-3-88 gallate caused the conformational changes in ?-lactalbumin, which were inducing ?-helix to ?-structures transition. 89 Not only the non-covalent bonds but also the covalent bonds also occur in a phenolic-protein complex. Sui 90

et al. (2018) demonstrated the conformational changes on soy proteins in the presence of anthocyanins. They used three-dimensional fluorescence and Fourier transform infrared spectroscopy to determine the interaction character. They reported that the covalent bonds occurred more abundant than non-covalent bonds in soy protein isolate-anthocyanin complex. The conformational changes on the secondary structure of soy protein isolate also detected as a decrease in ?-sheets and an increase in ?-turns and random coils.

# <sup>96</sup> 2 III. Interaction Effect on Nutritional and Antioxidant Quality <sup>97</sup> of Foods

The interaction of phenolics-proteins and starch may be one of the mostfundamental factors affecting the nutraceutical quality of the food products. There are several studies that is confirmed by earlier studies including phenolic-protein ?? Arts et The chemical structure of phenolic compounds is the enhancive reason for interaction with major food components. In other words, a hydroxyl group and carboxylic acids give rise to increasing of phenolic interaction with proteins, carbohydrates, lipids, and (Bravo, 1998; Alu'datt, Rababah, Ereifej, Brewer, & Alli, 2013; Escarpa & Gonzalez, 2001). As mentioned before, phenolic acids and proteins bring about two main

types of interactions (covalent or noncovalent) that result in two kinds of precipitation of proteins. First one is 104 multisite interactions, that is, several phenolics bind to one protein molecule and the second one is multidentate 105 interactions, that is, one phenolic binds to several protein sites or protein molecules. A study by Rawel, Meidtner, 106 and Kroll (2005) stated that the non-covalent binding of phenolic compounds does not affect n the secondary 107 108 structure of the proteins while may induce to distinct alteration in the tertiary structure of proteins. However, it has been indicated that the covalent binding of phenolics compounds may affect both tertiary and secondary 109 structures of proteins (Kroll et al., 2003). Moreover, these interactions may also end in changes in the thermal 110 stability, solubility, and digestibility of food proteins (Kroll et al., 2003;Ozdal et al., 2013). S. Wu et al. (2018) 111 have studied on 71 phenolic acids, and the derivatives of these phenolics were chosen to estimate the binding 112 affinity with ?-lactoglobulin. According to their study, the potential mechanisms for increased binding affinity 113 is the inclusion of the hydrogen bond in the interaction between the ?-lactoglobulin and phenolic acids. The 114 interaction between ?-lactoglobulin-phenolic acid has enhanced antioxidant activity when compared tothat of the 115 phenolic acids alone (S. Wu et al., 2018). This study gives perspective for understanding the relationship between 116 the chemical structure of phenolic acids and the affinity for ?-lactoglobulin.Volume XIX Issue I Version I Year 117 2019 (DDDD) L minerals 118

Most of the nutritional concern of phenolics focus on their side effects caused by the capability of phenolics to conjugate and precipitate protein, lipids, minerals and carbohydrates resulting in reducing food digestibility (Bravo, 1998). On the other hand, several studies indicated that regarding the bio-functional impact of the naturally occurring interactions between phenolics with food constituents such as antioxidant effects become inadequate (Bravo, Saura-Calixto, and Goni, 1992; Alu'datt, Rababah, Ereifej, Brewer, et al., 2013).

#### <sup>124</sup> 3 IV. Interaction Effect on Organoleptic Quality of Foods

Polyphenol compounds are currently present in multiform beverages and food products. Several properties of phenolic compounds are using in the food industry. Antioxidant properties of phenolic compounds have been utilized in the food industry to stabilize cysteine proteases and to prolong shelf-life of processed products (Howell, 2005). Also, phenolic compounds including anthocyanins, ?-carotene, riboflavin, and curcumin are using as natural food colorants in beverage products (Mortensen, 2006).

These compounds have poor solubility and limited chemical stability which is challenging as a natural colorant 130 131 in food products and beverages (Chung, C., Rojanasasithara, T., Mutilangi, W., & McClements, D. J., 2017). 132 On the other hand, phenolic-protein interaction is using to maintain the stability of phenolic-colorants. The interaction between protein and phenolic compounds is initially starting with a hydrophobic effect and is stabilized 133 by hydrogen bonding (Oh, Hoff, Armstrong, & Haff, 1980; Siebert, 1999). Thus, different research groups are 134 studying on improvement of the stability of phenolic-colorants by using phenolic-protein interaction phenomena. 135 Chung et al. (2015) proved that when heated whey protein isolates are added to model beverage systems 136 containing ascorbic acid, it is improved the color stability of anthocyanins in the beverage. Ascorbic acid has 137 a strong effect to tint of the anthocyanins in these systems. (Mercadante & Bobbio, 2007; Poei-Langston & 138 Wrolstad, 1981). Recently, Chung and co-workers studied beverage products containing ascorbic acid. They 139 examined the effect of polypeptides and amino acids on the color stability of anthocyanins. Also, the product 140 containing ascorbic acid and an amino acid is two times more convenient regarding to the average half-life of the 141 anthocyanin than alone ascorbic acid added the product (Chung C. et al., 2017). 142

Astringency sensation and bitterness are another important sensory attribute for wine. Interaction and precipitation of proline-rich proteins, especially, salivary proteins by tannins is estimated to be the general explanation of astringency onset ??Charlton et ??018) found evidence that the interaction of proline-rich proteins by tannins is disparate when the proteins are present simultaneously or alone. However, protein-phenolic interaction and co-protein interaction via tannins may be responsible and need to be more investigated.

In respect of food processing, there are only limited studies to confirm the effects of phenolic-protein 148 interactions. Indeed, for the food industry, the interaction is exploited as refining and clarification treatments to 149 improve haze stability (Cosme, Ricardo-da-Silva, & Laureano, 2008). There are few studies on improving textural 150 properties via the phenolic-protein interaction. Wu, Clifford, and Howell (2007) reported that the forming and 151 the gelling potential of egg albumin protein could be significantly increased by addition of the instant green tea. 152 In another study, green tea powder added to weat dough to enhance viscous/elastic modulus and the stability of 153 154 wheat dough. ??Wang et al., 2015). Having obtaining protein-phenolic interaction between fruit extract and soy 155 protein isolate in a glutenfree rice noodle, improved noodle quality, and the dough is achieved ?? enhanced via interaction of pea protein isolate with green tea extract to the supported both antioxidant activity and network. 156

Beyond that, presence of phenolic compounds at a nutritional daily intake had no adverse impact on protein digestibility. However, the high dose consumption of a phenolic extract may appear destructive to humans who consume low protein amount. Hence, consuming polyphenol supplements needs to be given to the benefit/risk balance with dedicated care (Dufour, C. et al., 2018).

# <sup>161</sup> 4 V. Interaction Effect on Bioaccessibility/Bioavailability of <sup>162</sup> Phenolics

To demonstrate the bioactivity of a compound on human health, it should have a sufficient bioaccessibility 163 and bioavailability. While bioaccessibility refers to the (%) undecomposed (bioactive) fraction of a compound 164 after gastro-intestinal digestion, bioavailability represents the (%) metabolized fraction of a bioactive compound. 165 Both bioaccessibility and bioavailability of phenolic compounds can be improved, impeded or unaltered when 166 co-delivered with other foods like rich in protein, carbohydrate, fat or fiber ?? There is currently a natural 167 affinity between polyphenols and proteins (Bandyopadhyay, Ghosh, & Ghosh, 2012). Thereby, protein-rich food 168 matrices can be stabilized and concentrate anthocyanins and also other phenolics Roopchand, Grace et al., 169 2012; Roopchand et al., 2013). Grape and blueberry polyphenol-enriched protein matrix have been studied for 170 oral administration, and the hypoglycemic effect obtained in mice indicates that the phenolic-protein complex 171 is bioactive. (Roopchand et al., 2013). In another study, it is reported that anthocyanins are conserved by 172 the sorption to defatted soy flour while transiting through the upper gastrointestinal tract by permitting huge 173 amounts to be gained to the colon (Ribnicky D. M. et al., 2014). Pineda-Vadilloet al. (2016) studied on protein-174 rich products enriched with grape extracts. As a result, they showed that the food matrix has no affect on the 175 176 antioxidant activity; while the antioxidant capacity was steady during the oral and gastric phases, it considerably increased during the intestinal phase of digestion. There is also another study, that supports the approach of 177 178 the noncovalently polyphenols-proteins interaction which are hydrolyzing during digestion, and there was no effect on the absorption of polyphenols and proteins (Budryn, G., & Nebesny, E., 2013). These studies indicated 179 that becoming complex with the protein matrix does not affect the bioaccessibility of anthocyanins/polyphenols 180 negatively (Budryn, G. & Nebesny, E. 2013; Ribnicky D. M. et al., 2014; Pineda-Vadillo et al., 2016). In the 181 previous nvivo models, it was demonstrated that the bioavailability of milk and coffee can reduce when they are 182 consumed together (Duarte & Farah, 2011). The different inferences among various studies may be the potential 183 of some phenolics to be complex with digestive enzymes and food matrix. The previous study of ??ohn et al. 184 (2002) confirmed that the activity of selected digestive enzymes such as trypsin and ?-amylase was a decline in the 185 case of protein-phenolic interaction. Thereby, the interaction has caused to the antioxidant activity of phenolic 186 compounds reduce as it cannot leave the protein-phenolic complex. However, the interaction of phenolic-protein 187 occur only some functional groups and the un-interacted part still can be show activity (Rohn, ??awel, & Kroll, 188 189 2002; Alminger, M. et al., 2014). Indeed, being complex with polyphenols is known to reduce protein digestibility via protecting proteins from enzyme degradation or through interaction with digestive enzymes. 190

## <sup>191</sup> 5 VI. Delivery Techniques based on Phenolic-Protein Interac-<sup>192</sup> tion

Phenolic compounds are highly unstable bioactive compounds due to exposure to degradation by light, soluble oxygen or enzymes. The basic goal of delivery techniques is to protect the phenolic compounds from adverse environmental conditions. Protein-based delivery techniques are one of the novel strategies from this area, and there are numerous studies about this subject. The delivered phenolic compounds, as well as the delivery system, is summarized in Table ??. These strategies can mainly divide into two groups; (i) protein-phenolic conjugates and (ii) protein-based nanoparticles.

#### <sup>199</sup> 6 a) Protein-phenolic conjugates

Protein-phenolic conjugates can be used to enhance antioxidant activity of a protein as well as its stability 200 (Frazier et al., 2010; Wu et al., 2011; El-Maksoud et al., 2018) or to reduce the degradation level of delivered 201 phenolic compounds. Since the non-covalent interactions are reversible, the stable structure cannot be obtained 202 by non-covalent bonds. Besides, these types of interactions lead to alterations on protein and phenolic compound 203 structure, hence their functionality and nutritional value are changed (Mehanna et al., 2014;Ozdal et al., 2013). 204 Conversely, conjugates are constructed with covalent bonds. Hence stable structure can be obtained. El-Maksoud 205 et al. phenolic compound that they used in conjugate was caffeic acid, and ?-lactoglobulin was selected for protein 206 part of conjugates. They reported that the conjugates showed better water solubility than native ?-lactoglobulin 207 and non-covalently bond ?-lactoglobulin-caffeic acid complex. Moreover, the thermal stability of ?lactoglobulin 208 significantly was increased with this conjugate. 209

210 In the view of nutraceutical delivery aspect, protein-phenolic conjugates offer several advantages. 211 vestigated the physicochemical properties of ?-carotene emulsions stabilized via chlorogenic acid-lactoferrin-212 glucose/polydextrose conjugates. They indicated that the produced conjugate offered better emulsifying prop-213 erties such as the physicochemical stability of ?-carotene emulsions can be conserved during the freeze-thaw 214 treatment. Besides, chemical stability of ?-carotene in the emulsions against ultraviolet light exposure can enhanced by the conjugate. Therefore, they suggested that the conjugates containing protein, polyphenol and 215 carbohydrates could be a smart building block for delivery systems . In another study, the chemical stability 216 of curcumin to degradation at physiological pH and of resveratrol to degradation under ultraviolet irradiation 217 conditions was obtained by zeinepigallocatechin gallate conjugates (Liu et al., 2018). 218

#### <sup>219</sup> 7 b) Protein-based nanoparticles

Zein and gliadin are the prolamine-type proteins which generally occur in in cereals such as corn and wheat, 220 respectively. The four major components of zein are ?, ?, ? and ?-zein (Hu & McClements, 2015). Since both 221 zein and gliadin contain the high amount of nonpolar amino acids in their primary structure, they are soluble 222 in aqueous ethanol solution (60-90%), but insoluble in water ??Rombouts et al., 2009; ??hukla & Cheryan, 223 2001). Because of their highly hydrophobic nature, these proteins can be easily converted into spherical colloidal 224 nanoparticles, which are effective delivery agents for phenolic compounds (Chen, Zheng, McClements, & Xiao, 225 2014). Previous studies reported that protein-based nanoparticles are suitable and effective delivery agents for 226 different phenolic compounds (Table ??). During the formation of proteinbased nanoparticles with phenolics, 227 non-covalent interactions such as electrostatic interaction, hydrogen bonding, and hydrophobic interactions were 228 involved in the structure (Dai et al., 2018). On the other hand, these interactions mainly depend on the type 229 of protein and phenolic compounds. Joye et al. (2015) studied on binding ability of resveratrol to the zein and 230 gliadin. They assumed that hydrogen bonds are the main force that contributed to the interaction between 231 resveratrol and zein. However, the hydrophobic interactions constructed the resveratrol-gliadin interaction. In 232 another study about curcumin delivery by zein-nanoparticles, are indicated that hydrogen bonds between the 233 phenolic hydroxyl groups in curcumin and the carbonyl group in amide bonds in zein were attributed to the 234 formation of proteinbased nanoparticle with polyphenol (Dai et al., 2017;Sun et al., 2017). These nanoparticles 235 are not only protecting the related phenolic compounds form adverse environmental conditions, beyond that they 236 support the controlled released of the phenolics. Liang et al. (2017) reported that the controlled release property 237 of epigallocatechin gallate was improved by zein/ chitosan nanoparticles and according to Sun et al. (2017), 238 controlled release of curcumin during in vitro digestion, can be obtained by zein-shellac composite colloidal 239 particles. In another study, in vitro release of curcumin as well as its stability are improved by zein nanoparticles 240 (Dai et al., 2018). The authors suggested that curcumin might bind to zein in tyrosine residue. Since, the aromatic 241 242 side groups and double bonds in zein molecules could absorb UV light (Luo et al., 2013), zein nanoparticles 243 enhance the stability of curcumin against UV light. The zein-lecithin composite nanoparticles also improved the 244 stability of curcumin against UV irradiation, high ionic strength and thermal treatment (Dai et al., 2017).

## 245 8 VII. Conclusion

The interaction between phenolic compounds and proteins is an important phenomena since it affects the 246 functionality, biological activity and nutritional quality of protein and phenolics. The interaction is contributed 247 with both non-covalent and covalent bonds that depend on the type of protein and phenolics as well as the 248 environmental conditions. Depending on the bonding type, there occur conformational changes in protein 249 250 structure. There are several techniques for determining the bounding type such as spectroscopic methods, 251 microscopic methods, thermodynamic methods, bioinformatics methods, electrophoretic and chromatographic methods. The delivery of phenolics in the desired system can be done by novel agents which are constructed with 252 proteins. Indeed, the bonding type is important to select the novel delivery strategies. If the protein functionality 253 is important in delivery system, then the covalent bonds are crucial to eliminating the structural changes. But 254 if the controlled released of phenolic is desired, the non-covalent bonds are wanted. 255

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<sup>&</sup>lt;sup>4</sup>Phenolic-Protein Interaction: Effects on Functional Properties of Phenolics and Advantages on Phenolic Delivery Platform Development

### 1

Polyphenol	Protein	Determination Method	Type of Interaction		
Green tea	Milk pro- teins	Fluorescent probe	Hydrophobic interactions		Protein surface h
flavanoids		binding method	between catechin and ?-		was
		Isothermal titration calorimetry	casein.		hydrophobic bind milk proteins and
Pelargonidin	Dairy pro- teins:	Fluorescence	Hydrophobic interactions		the structural cor
		spectroscopy	between pelargonidin-?-		the milk proteins
	?- lactoglobulin		lactoglobulin		binding process
	Caseinat	;e	Hydrogen between caseinate	bonding pelargonidin-	
Chlorogenic acid	?- lactogloł	Fluorescence oulin	Hydrogen bonding and		The secondary st
Ferulic acid	spectroscopy		Van der Waals interactions between ?-		lactoglobulin

Figure 1: Table 1 :

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   .
- $_{260} \quad [{\rm Wang \ et \ al.}] \ , {\rm Q \ Wang \ , \ Y \ Li} \ , {\rm F \ Sun \ , \ X \ Li} \ , {\rm P \ Wang \ , \ J \ Sun \ .}$
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