Colourimetric Assay of Atomoxetine Hydrochloride by Simple Aurum Coupling Reaction in Bulk and Tablet Dosage Form

By B. Mohammed Ishaq, Hindustan Abdul Ahad, Shaik Muneer & S. Praveena

Balaji College of Pharmacy, India

Abstract- Simple, rapid and sensitive spectrophotometric procedure was developed for the analysis of atomoxetine hydrochloride (ATH) in pure form as well as in pharmaceutical formulations. The method was based on the reaction of ATH with gold (III) chloride in the pH range 3.5-4.5 forming violet colored complex solution, showing absorption maxima at 550 nm. The linear plot indicates that Beer’s law is obeyed in the range of 5 – 80 μg/ml of atomoxetine hydrochloride. The molar absorptivity and Sandell’s sensitivity are 3.77 x 103 M and 0.0774 μg cm^{-2} respectively. The standard deviation of the method for ten determinations ATH is 9.9827 x 103. The correlation coefficient (r2) of the experimental data of the calibration plot is 0.9997. The effective range of concentration for accurate determination of ATH as ascertained from Ringbom’s plot and it is 10 – 80 μg/ml.

Keywords: atomoxetine hydrochloride, spectrophotometric, ringbom’s plot, pharmaceutical formulations.

GJMR-B Classification : NLMC Code: QV 786, WA 730

© 2013, B. Mohammed Ishaq, Hindustan Abdul Ahad, Shaik Muneer & S. Praveena. This is a research/review paper, distributed under the terms of the Creative Commons Attribution-Noncommercial 3.0 Unported License http://creativecommons.org/licenses/by-nc/3.0/), permitting all non-commercial use, distribution, and reproduction in any medium, provided the original work is properly cited.
Colourimetric Assay of Atomoxetine Hydrochloride by Simple Aurum Coupling Reaction in Bulk and Tablet Dosage Form

B. Mohammed Ishaq α, Hindustan Abdul Ahad α, Shaik Muneer α & S. Praveena α

Abstract—Simple, rapid and sensitive spectrophotometric procedure was developed for the analysis of atomoxetine hydrochloride (ATH) in pure form as well as in pharmaceutical formulations. The method was based on the reaction of ATH with gold (III) chloride in the pH range 3.5–4.5 forming violet colored complex solution, showing absorption maxima at 550 nm. The linear plot indicates that Beer’s law is obeyed in the range of 5 – 80 μg/ml of atomoxetine hydrochloride. The molar absorptivity and Sandell’s sensitivity are 3.77 x 10^3 M and 0.0774 µg cm-2 respectively. The standard deviation of the method for ten determinations ATH is 9.9827 x 10^3. The correlation coefficient (r^2) of the experimental data of the calibration plot is 0.9997. The effective range of concentration for accurate determination of ATH as ascertained from Ringbom’s plot and it is 10 – 80 μg/ml.

Keywords: atomoxetine hydrochloride, spectrophotometric, ringbom’s plot, pharmaceutical formulations.

I. Introduction

(-)-N-Methyl-3-phenyl-3-(o-tolyloxy)-propylamine hydrochloride is atomoxetine hydrochloride (ATH). The molecular formula is C17H21NO•HCl. Its molecular weight is 291.82. The structure of automoxetin is as follows

![Figure 1: Chemical Structure of atomoxetine hydrochloride](image)

It is practically a white solid and has a solubility of 27.8 mg ml-1 in water. It is the first nonstimulant drug approved by the FDA for the treatment of attention-deficit hyperactivity disorder (ADHD) in children, adolescents and adults. ADHD is the most common neurobehavioral disorder among children with an estimated worldwide prevalence of 8–12%.

ATH is not official in IP, BP, USP and EP. AMX is available commercially as capsules under brand name Straterra, Eli Lilly and company capsules. AMX capsules are intended for oral administration only. The capsules are available with strengths of 10, 18, 25, 40, 60 and 80 mg of ATH base. The capsules also contain pregelatinized starch and dimethicone.

To the best of our knowledge, there is no work in the literature reported about the colourimetric method for the analysis of ATH in pharmaceutical formulations. Hence the author has made an attempt to develop simple and sensitive spectrophotometric method for the estimation of ATH in bulk drugs and in pharmaceutical formulations. The method was based on the reaction with gold(III) to form a violet coloured complex in the pH range 3.5 – 4.5.

II. Materials and Methods

All chemicals used were of analytical reagent grade and double distilled water was used for preparing the reagent solutions. ATH was obtained from Dr. Reddy’s labs Hyderabad. Stock solution of ATH was freshly prepared by dissolving 100mg of ATH in 100mL of distilled water and then this was further diluted with distilled water so as to obtain working standard solution of 100 μg/mL.

a) Apparatus

All spectral and absorbance measurements were made on a Shimadzu UV-Visible digital Spectrophotometer (UV-160A) with 10mm matched quartz cells.

b) Pharmaceutical dosage form

Tablet label claim of 10 mg ATH (Straterra, Eli Lilly) was procured from local market.

c) Preparation of Atomoxetine hydrochloride solution

100 mg of ATH was weighed accurately and transferred into a 100mL standard flask, dissolved and made up to the mark with methanol. This solution is diluted as required.

Authors α α p О: Department of Pharmaceutical Analysis, Balaji College of Pharmacy, Anantapur, A.P India.
e-mail: bmdishaq@yahoo.com
d) Preparation of Gold (III) solution

1gm of chloroauric acid (Johnson Mathews, materials technology, U.K.) is dissolved in distilled water after adding few drops dilute HCl. The solution is made upto the mark in 100 ml volumetric flask. The gold content of the solution is determined by rhodamine B method8. The working solutions are prepared by diluting the stock solution.

III. Experimental Method

a) Determination of Atomoxetine hydrochloride

To explore the possibility of employing the colour reaction for the determination of ATH, the absorbance of the experimental solution containing different amounts of ATH, keeping the Au (III) concentration constant is measured in the wavelength range 400 – 700 nm.

b) Determination of gold (III)

To explore the possibility of employing the colour reaction for the determination of gold(III) in trace level, the absorbance of the experimental solutions containing different amounts of gold(III), keeping the ATH concentration in excess, is measured in the wavelength range 400 – 700 nm.

c) Assay of Pharmaceutical dosage form of Atomoxetine hydrochloride

The present method for the determination Atomoxetine hydrochloride is applied for its determination in a pharmaceutical sample. A known aliquot of pharmaceutical sample solution of atomoxetine hydrochloride is added to a 10 ml volumetric flask containing 5ml of buffer solution of pH 4.0 and 0.5 ml of gold(III) (5.0 x 10-3M) solution 1.5 ml of 2% SDS solution. The contents are made upto the mark with distilled water. After heating for 60 minutes at 650C and cooling the solution to room temperature. The absorbance of the resulting solution is measured at 550 nm against the buffer blank. The amount of atomoxetine hydrochloride is computed from the predetermined calibration plot at 550 nm.

IV. Results and Discussion

The spectra presented in fig 2 show that the complex has an absorption maximum at 550 nm. Neither gold (III) nor atomoxetine hydrochloride have absorbance at 550 nm. Hence, analytical studies are made at 550 nm. However, in presence of excess atomoxetine hydrochloride the complex shows maximum absorbance at 550 nm.

Figure 2: Absorption spectra of AMX – Au(III) system

a. AMX Vs buffer blank
b. Au(III) Vs buffer blank
c. AMX – Au(III) Vs buffer blank

\[ [\text{Au(III)}] = 5.0 \times 10^{-3} \; \text{M} ; \; [\text{AMX}] = 3.42 \times 10^{-3} \text{M} \]

a) Linearity

A plot of absorbance Vs amount of atomoxetine hydrochloride is presented in fig. 3 the straight line plot obtained obeys the equation \( A = 0.129C - 0.0009 \). The linear plot indicates that Beer’s law is obeyed in the range of 5.0 – 80.0 \( \mu \text{g/ml} \) of atomoxetine hydrochloride.

The molar absorptivity and Sandell’s sensitivity are \( 3.77 \times 10^3 \, \text{l mol}^{-1} \, \text{cm}^{-1} \) and \( 0.0774 \, \mu \text{g/cm}^2 \) respectively. The standard deviation of the method for ten determinations of 10 \( \mu \text{g/ml} \) of atomoxetine hydrochloride is 9.9826 x 10-4. The correlation coefficient (\( \gamma \)) of the experimental data of the calibration plot is 0.9997. The effective range of concentration for accurate determination of atomoxetine hydrochloride as ascertained from Ringbom’s plot and it is 10.0 – 70.0 \( \mu \text{g/ml} \).

Figure 3: Absorbance Vs amount of AMX (\( \mu \text{g/ml} \))

\[ [\text{Au(III)}] = 5.0 \times 10^{-3} \text{M} ; \; \text{pH} = 4.0 ; \; \lambda = 550 \text{ nm} \]

b) Effect of excipients

Various amounts of excipients that are generally associated with the atomoxetine hydrochloride in its pharmaceutical formulations are added to a fixed amount of atomoxetine hydrochloride (10 \( \mu \text{g/ml} \)) solution and the absorbance measurements are carried.
out under optimal conditions. The concentration (µg/ml) at which various ions do not cause an error of more than ± 4% in the absorbance is taken as the tolerance limit and the results are given in table 1.

**Table 1**: Tolerance limit of excipients Amount of AMX = 10 µg/ml ; pH = 4.0

<table>
<thead>
<tr>
<th>Excipient</th>
<th>Tolerance limit (µg/ml)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Fructose</td>
<td>1257</td>
</tr>
<tr>
<td>Glucose</td>
<td>901</td>
</tr>
<tr>
<td>Sucrose</td>
<td>1369</td>
</tr>
<tr>
<td>Lactose</td>
<td>1704</td>
</tr>
<tr>
<td>Gelatin</td>
<td>1811</td>
</tr>
<tr>
<td>Starch</td>
<td>1424</td>
</tr>
<tr>
<td>Sodium Alginate</td>
<td>1324</td>
</tr>
<tr>
<td>Boric acid</td>
<td>1891</td>
</tr>
<tr>
<td>Magnesium stearate</td>
<td>1576</td>
</tr>
</tbody>
</table>

The data in table 1 indicate that the excipients that are associated with atomoxetine hydrochloride do not interfere even in large quantities in the determination of atomoxetine hydrochloride making the method highly selective and direct.

**Table 3**: Optimal and regression characteristics, precission and accuracy of the proposed method for atomoxetine hydrochloride and gold(III)

<table>
<thead>
<tr>
<th>Parameter</th>
<th>Atomoxetine hydrochloride</th>
<th>Gold(III)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Analytical wavelength (nm)</td>
<td>550</td>
<td>530</td>
</tr>
<tr>
<td>Beer’s law limits (µg/ml)</td>
<td>5.0 - 80.0</td>
<td>9.84 – 157.42</td>
</tr>
<tr>
<td>Limits of detection (µg/ml)</td>
<td>2.2837</td>
<td>10.3723</td>
</tr>
<tr>
<td>Limits of quantization (µg/ml)</td>
<td>7.6170</td>
<td>34.7446</td>
</tr>
<tr>
<td>Molar absorptivity (µmol⁻¹·cm⁻¹)</td>
<td>3.77 x 10⁻³M</td>
<td>0.75 x 10⁻³M</td>
</tr>
<tr>
<td>Sandell’s sensitivity (µg·cm⁻²)</td>
<td>0.0774</td>
<td>0.2625</td>
</tr>
<tr>
<td>Regression equation (y = a + bx)</td>
<td></td>
<td></td>
</tr>
<tr>
<td>Slope (b)</td>
<td>0.0129</td>
<td>0.0047</td>
</tr>
<tr>
<td>Intercept (a)</td>
<td>-0.0009</td>
<td>0.0104</td>
</tr>
<tr>
<td>Correlation coefficient (γ)</td>
<td>0.9997</td>
<td>0.9981</td>
</tr>
<tr>
<td>Standard Deviation (SD)</td>
<td>9.9827 x 10⁻³</td>
<td>0.0163</td>
</tr>
</tbody>
</table>

V. Conclusions

Atomoxetine hydrochloride reacts with gold(III) to form stable violet coloured 1 : 1 complex at pH 4.0. Spectrophotometric and derivative spectrophotometric methods are developed based on this reaction. They are sensitive for the assay of both atomoxetine hydrochloride and gold(III). The tolerance limit of the excipients and the foreign ions in derivative methods is found to be generally 10 – 20% greater than that of the zero order method. The present spectrophotometric methods are direct, simple and highly selective for the determination of gold(III) or atomoxetine hydrochloride. Further, the methods can easily be employed by ordinary clinical laboratories as the methods can be carried out using a simple colorimeter.

**References Références Referencias**