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1	Implant Surface Micro-Design
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6 Abstract

The application of implants for dental and orthopedic surgery has increased rapidly within the 7 past few decades. In craniomaxillofacial surgery, different implant systems have been applied, 8 for example, for dental and bone replacement or osteosynthesis plates and screws. These 9 implants may be made of pure titanium or a titanium alloy, usually titanium aluminum-10 vanadium (Ti-6Al-4V). The surface can be turned or Machined or a coating may cover the 11 metal base. The reason for treating the implant surface is to obtain maximum boneimplant 12 contact and bone-implant stability and to shorten the healing time for earlier loading. The 13 crucial aspect of pure titanium implants is the development of titanium oxide on the surface. 14 This oxide and other known coatings for implant material do not have high wear resistance. 15 This article thus aims to review Implant Surface Micro-design its rationale, various surface?s 16 physical and chemical properties, different types of implant surface treatments, optimum 17 roughness of oxidized implants and controversies associated with various implant 18 topographies. The recent advances like nanotechnology are also included. 19

21 Index terms— implants, implant topography, implant surface micro-design, implant surface treatments.

²² 1 Introduction

he success and predictability of osseointegrated dental implants have forever changed the philosophy and practice 23 of dentistry and, perhaps more than any other specialty, Prosthodontics has changed dramatically. In the late 24 25 1950's, Per-Ingvar Branemark, a Swedish professor in anatomy studying blood circulation in bone and marrow, developed through a serendipitous finding in the history of medicine: he predictably achieved an intimate bone-26 to implant apposition that offered sufficient strength to cope with load transfer. He called the phenomenon 27 "osseointegration".. Since that time, millions of patients have been treated worldwide using this technique. The 28 implants used sometimes had different geometries and surface characteristics. A key element in the reaction 29 of hard and soft tissues to an implant involves the implant's surface characteristics, that is, the chemical and 30 physical properties. Quest continued for a material with a surface property which enhances bone apposition at 31 the implant surface in an osteoconductive manner. The quest was for a biocompatible if not bioactive surfaces, 32 achieved through additive or subtractive process. Titanium, preferably commercially pure titanium, became the 33 standard for endosseous implants. Actually titanium is a very reactive material that would not become integrated 34 with tissues. However, its instantaneous surface oxidation creates a passivation layer of titanium oxides, which 35 have ceramic-like properties, making it very compatible with tissues. 36

37 **2 II.**

20

³⁸ 3 Rationale for a Dynamic Implant Surface

Oral implant is an alloplastic material or device that is surgically placed in to the oral tissue beneath the mucosal or periosteal layer or within the bone for functional, therapeutic, or esthetic purposes 1. More needs to be known about the optimal situation of the connection between an artificial material and the tissues-what type of

6 A) SURFACE MICROSTRUCTURE

42 material that gives the best tissue response and what type of surface is preferred by the bone cells or the cells in 43 the soft tissue. If this is known, the response of the bone or soft tissue can be predicted when the implants are

installed into the jaws. There is some information and understanding of the effect of design and toxicology of the

45 implants, surgery techniques, effect of movement of the implant during the healing period and biodegradation.

46 Understanding is lacking, however, of the relationship between the events that occur at the implant surface and

the effect the implant material has in the tissue and the biocompatibility of the material 2.
a) The bone-implant interface Bone tissue is a living organ, which can be described as a natural composite

composed of an organic matrix strengthened by an inorganic calcium phosphate (CaP) phase. The extracellular organic matrix (ECM) of bone consists of 90% collagenous proteins and 10% non-collagenous proteins. Regarding the inorganic component, the most abundant mineral phase in human bone is carbonate rich hydroxyapatite (with

 $_{52}$ $\,$ a carbonate content between 4% and 8%) 3 .

⁵³When an implant is installed in a jaw, a series of reactions take place on the implant surface. The implant ⁵⁴ is exposed to a series of different ions, to polysaccharides, carbohydrates and proteins as well as to such cells ⁵⁵ as chondroblasts, fibroblasts and osteob-lasts that react with the surface (Figure ??1 and 1a) 2,3. The initial

reactions between the tissue constituents and the implant surface govern the further reactions and determine the biological activity of the surface and the further cell responses to the surface. This tissue response depends on

the nature of the surface and its chemical properties, which influences the nature of the subsequent composition of the protein film that adsorbs onto the material [4][5][6][7]. this further strongly influences the cell responses on the surfaces.

Figure ?? : After implantation, the biomaterial is exposed to a series of different tissue constituents that react with the surface. The type of reaction that occurs probably influences the further cell reactions and finally the tissuebiomaterial connection. Ig: immunoglobulins Figure ??(a) : Schematic representation of events consecutively taking place at the titanium surface after implantation into living bone tissue. Water binds to the surface, followed by incorporation of hydrated ions, adsorption and desorption of proteins, eventually leading to cell attachment. After differentiation, mature osteoblasts produce the extracellular matrix (ECM) b) Osseointegration versus Osseo-coalescence

The term osseointegration largely refers to the physical integration or mechanical fixation of an implant in bone. The interlocking provides mechanical resistance to forces such as shear experienced in "pull-out" and "torque-out".

70 With purely physical interaction, however, the interface would not be able to withstand even moderate tensile

71 forces. The term osseocoalescence has been proposed to refer specifically to chemical integration of implants in

⁷² bone tissue. The term applies to surface reactive materials, such as calcium phosphates and bioactive glasses,

 73 which undergo reactions that lead to chemical bonding between bone and biomaterial. With these materials, the

74 tissues effectively coalesce with the implant 8.

75 **4 III.**

76 5 Physical Properties

Several authors have discussed the dimension of the ideal roughness that would provide increased retention and an improved bone response. The roughness can be considered on different levels: macrostructural, microstructural and ultrastructural, and roughness on these different levels probably has different effects on the living tissues. It has been established in the literature based on several studies that, to gain complete growth of bone into a material's irregularities, these need to be at least 100 µm in size. Growth of bone into cavities or pores of this size will give a mechanical interlocking of the material with bone. This was demonstrated by Bobyn et al. in studying cobaltbased alloys with pore sizes of 50-400 µm 9 , Bone ingrowth was also observed by Clemow et al.

when this group studied porous coated Ti&l,V femoral implants with pore sizes ranging from 175 to 235 μ m 10 .

6 a) Surface Microstructure

This can vary considerably depending on the surface treatment of the implant. Variation of the surface microstructure has been reported to influence the stress distribution, retention of the implants in bone and cell responses to the implant surface. The implants with rough surfaces have improved bone response, with bone trabeculae growing in a perpendicular direction to the implant surface. An improved retention in bone has also previously been reported after implantation of rough-surfaced implants 2.

Surface roughness on a smaller scale was, however, found to be important for integration of the bone with 91 the implant surface 11. Although surface roughness on a micrometer scale gives some retention due to bone 92 93 in growth, in vitro cell studies indicate that this property of the surface influences the function of the cells, 94 the matrix deposition and the mineralization 12. Cells seem to be sensitive to microtopography and appear 95 to be able to use the morphology of the material for orientation and migration 13. The maturation of the 96 cells also affects the response to the surface roughness, which is in agreement with earlier observations that indicated that chondrocytes are affected differently by local factors such as vitamin D and transforming growth 97 factor p depending on the stages of maturation of the cells 14,15 . Microtopography may therefore be one factor 98 that influences the differentiation of mesenchymal cells into fibroblasts, chondrocytes or osteoblasts. Based on 99 these studies, it can be hypothesized that osteogenesis may be favored by vascular in growth, whereas a limited 100 vascular in growth may induce chondrogenesis. Figures: 2 and The ideal surface roughness for bone implants on a 101

micrometer scale probably depends on the distribution of cortical or cancellous bone and on the level of loading to 102 the implants. 2 The rugofile bone cells recognizes the surface prepared by the course particle, as a smooth surface, 103 whereas the 25-pm particles creates a rough surface that is identified by the osteoblasts Osteoblasts respond to 104 105 microarchitectural features of their substrate. On smooth surfaces (tissue culture plastic, tissue culture glass, and titanium), the cells attach and proliferate but they exhibit relatively low expression of differentiation markers 106 in monolayer cultures, even when confluent. When grown on microrough Ti surfaces with an average roughness 107 of 4-7 ?m, proliferation is reduced but differentiation is enhanced and in some cases, as it is synergistic with 108 the effects of surface microtopography. In addition, cells on microrough Ti substrates form hydroxyapatite in 109 a manner that is more typical of bone than do cells cultured on smooth surfaces. Osteoblasts also respond to 110 growth factors and cytokines in a surface-dependent manner. On rougher surfaces, the effects of regulatory factors 111 like 1?, 25(OH) 2 D 3 or 17?-estradiol are enhanced. When osteoblasts are grown on surfaces with chemistries 112 or micro architectures that reduce cell attachment and proliferation, and enhance differentiation, the cells tend 113 to increase production of factors like TGF ?1 that promote osteogenesis while decreasing osteoclastic activity. 114 Thus, on microrough Ti surface, osteoblasts create a microenvironment conducive to new bone formation 16. 115 Figure: Although micro-roughness seems to be an important characteristic for tissue response to biomaterials, 116 there are also observations that indicate a biological response to irregularities on the nanometer level. Larsson 117 118 et al. studied the biological effect of changing the oxide thickness of titanium implants from an electropolished 119 level, to thick oxide layers formed by anodization. By this treatment the surface changes from an amorphous 120 metal surface with a noncrystalline oxide to a polycrystalline metal surface with a crystalline oxide layer 17.

Analysis of these surfaces at a high resolution level demonstrated that the new surface was heterogeneous with mainly smooth areas of thick oxide but separated with porous regions on a nanometer level. This observation of an increased roughness after anodization of titanium was in line with earlier transmission electron microscopic studies demonstrating increased pore sizes with increased oxide thickness 18.

Implants with this thick, heterogeneous oxide seemed to have a slightly improved response in bone, particularly in the first weeks after implantation. c) Smooth versus Rough Surfaces Surface quality of an oral implant can be subdivided in to mechanical, topographic, and physicchemical properties 19. Surface topography is characteristic of the preparation process. Variations in the roughness and porosity can be categorized in function of the surfacing process. The current state of information regarding implant surface topography has provided clinicians with confusing options. Machined implants are not smooth, and not all rough implant surfaces are equivalent. Surfaces often are identified by the method of manufacture and not the resultant surface.

Rough surfaces can be divided into three levels depending on the scale of the features: macro-, microand nano-sized topologies. The macro level is defined for topographical features as being in the range of millimeters to tens of microns 20. This scale is directly related to implant geometry, with threaded screw and macro porous surface treatments giving surface roughness of more than 10?m. Numerous reports have shown that both the early fixation and long-term mechanical stability of the prosthesis can be improved by a high roughness profile compared to smooth surfaces 21.

The high roughness resulted in mechanical interlocking between the implant surface and bone on growth. However, a major risk with high surface roughness may be an increase in peri-implantitis as well as an increase in ionic leakage. A moderate roughness of 1-2?m may limit these two parameters 22. The microtopographic profile of dental implants is defined for surface roughness as being in the range of 1-10?m. IV.

¹⁴³ 7 Chemical Properties a) The surface chemistry of the implants

144 The chemical properties of the biomaterial surface play an important role for the tissue responses elicited by the material. This is at least one main reason why the tissues responds differently to different materials. 2 A 145 material with a surface that is accepted by the tissue seems to exhibit improved integration with bone, either 146 due to passive growth, leading to a tight connection between implants and bone, or by stimulation that probably 147 leads to a bone-implant bonding. This is probably the case with the two main materials used in dental implants, 148 hydroxyapatite and J titanium. 2 The calcified parts of the bone consists of hydroxyapatite (or rather carbonated 149 apatite), and introducing this substance as an implant material often gives favorable responses in the bone. 23 150 The biological effects of modifying the biomaterial surface have also been elaborated [24][25]. In an attempt 151 to study the effect of the oxide layer of titanium on calcium-phosphate precipitation, titanium-dioxide (TiO2) 152 and powder of oxidized and nonoxidized titanium were introduced into an in vitro nucleation test system 24 153 . In this system they found that titanium powder enhances calcium phosphate nucleation only after prolonged 154 preincubation in an aqueous buffer, or after autoclaving. These treatments enhance the growth of the oxide layer. 155 156 This observation indicated that the oxide content, or structure, is required for titanium to act as a nucleation 157 substrate. Even more effective nucleation was observed when pure TiO2 was used as a nucleation substrate. 158 The nucleation capacity and formation of calcium phosphate precipitates is related to the biocompatibility of titanium, and enhanced nucleation capacity may indicate improved biocompatibility. 2 The biological activity 159 of the TiO 2 probably also influences the protein adsorption to titanium. In an in vitro study, serum proteins 160 seemed to adsorb to titanium dioxide by the same mechanisms as to hydroxyapatite through calcium binding 161 24. The surface characteristics of TiO2 probably change from an anionic to a cationic state by the adsorption 162

of calcium to the surface. This will subsequently increase its ability to adsorb acidic macromolecules, such as albumin, a property demonstrated for hydroxyapatite [26][27].

Fluoride ions have documented activity in bone. This element is known to form fluoridated hydroxyapatite or fluorapatite with improved crystallinity and better resistance to dissolution than hydroxyapatite 28. Fluoride also enhances the incorporation of newly formed collagen into the bone matrix and increases the rate of seeding of apatite crystals as well as increasing trabecular bone density and stimulating osteoprogenitor cells number in vitro ??9,30.

¹⁷⁰ 8 Diferent Types of Implant Surface Treatments

The desired implant surface can be achieved by addition of material over the surface, removal of material from the surface or modification of the surface material. Some of the examples are: I. Addition of material -Titanium plasma spray (TPS, TiO2); coating with hydroxyapatite (HA).

II. Removal of material -Particle jets and/or acid etching III. Modification of material -The implant surface can
be modified without either adding or removing material. (Electron beam, thermal treatment, laser treatment,
and ion implantation) a) Addition of Material (Additive Methods)

177 Chemical substances are successfully added over the surface. Some of the materials used for this purpose 178 include: VI.

¹⁷⁹ 9 Additive Methods of Surface Treatment a) Hydroxyapatite ¹⁸⁰ coating

Hydroxyapatite is a calcium phosphate ceramic that is an osteophilic, osteoconductive, bioactive coating, which is totally biocompatible and becomes an integral part of living bone tissue. Hydroxyapatites and tricalcium phosphates have an excellent grade of acceptance, and these materials may be more rapidly incorporated in bone than commercially pure titanium. Hydroxyapatite coating over titanium has enjoyed a rapid growth because of its inherent biomaterial properties that some consider an advantage over uncoated surgical titanium. Hydroxyapatite (HA) coating has become popular for load bearing dental implants because it elicits a faster bony adaptation, absence of fibrous tissue seams, firmer implant bone attachment, reduced healing time, increased tolerance of

¹⁸⁸ surgical inaccuracies, and inhibition of ion release 31.

The first clinical use of hydroxyapatite (HA) as a coating on dental implants began in February 1984, with the results showing many benefits over the no coated implants31. Later, many researchers conducted studies and obtained promising results.

Contemporary plasma-sprayed hydroxyapatite (HA) coatings with high crystalline content are much more resistant to in vivo degradation than HA coatings of a decade ago but reportedly exhibit reduced wettability, which could potentially negatively affect tissue adhesion and long-term clinical outcome.

Bone morphogenetic proteins (BMP s) play a crucial role in cell ingrowth and differentiation in a variety of cell types, including osteoblasts 32. Because of their beneficial effects, BMP s have been used to accelerate healing after implant placement. Apatite is considered a suitable carrier of BMP-2 33 and the incorporation of BMP-2 into the apatite layer of a titanium implant may enhance its osteoinductive properties. i.

¹⁹⁹ 10 Methods of HA Coating

Conventional plasma spraying, flame spraying, and chemical techniques have all been investigated as techniques 200 for producing a thinner HA coating on a metal substrate. The bond formed between HA coatings and the metallic 201 substrate by the spraying method, formed primarily through mechanical interlocking, is not strong enough. 202 Additionally, the spraying method is unsatisfactory for applying a thinner, uniform HA coating on implants 203 because of their complicated shapes. On the other hand, electrochemical methods, electrop-Ohoretic techniques 204 in particular, seem attractive for forming HA coatings on titanium implants with complicated shapes. However, 205 the bond between the coating and the metal substrate is significantly weak. Magnetron sputter coating and Ion 206 Beam sputtering techniques for coating HA on implant surfaces have been tried with varying rates of success. 207 Hydroxyapatite can be coated by plasma spraying. In this technique, powdered crystalline hydroxyapatite is 208 introduced and melted by the hot, high velocity region of a plasma gun and propelled onto the metal implant 209 as a partially melted ceramic. b) Plasma Spraved Titanium Hahn and Palich (1970) first developed titanium 210 surfaces by plasma spray techniques and reported an enhanced bone ingrowth in those implants. The plasma 211 sprayed titanium surfaces exhibit a porous surface with macro irregularities 34. 212

²¹³ 11 i. Macro-irregularities

Macro-irregularities in an implant include macroscopic threads, fenestrations, pores, grooves, steps, threads, or other surface irregularities that are visible. The idea is to create mechanical interlocking between implant and bone at the macro level.

ii. Method of Plasma spraying Powdered Titanium is melted at a temperature of 15,000 degrees and is sprayed on to the surfaces of the implant at a very high velocity of 600 m/sec through argon plasma associated with a nozzle. The diameters of the sprayed particles are around .04 to .05mm thickness. When observed microscopically the coatings show round or irregular pores that are interconnected with each other. The surface of the implants where they condense and fuse together, forming a film about 30 ?m thick. The thickness must reach 40-50 ?m to be uniform. The resulting TPS coating has an average roughness of around 7 ?m, which increases the surface area of the implant. c) Anodic Spark Deposition Anodic spark deposition techniques have been effectively applied to achieve a microporous morphology on metals. Recently, a new electrochemical process has been developed to

²²⁵ improve further the mineralization potential, mechanical stability, and corrosion resistance of the ceramic coating

obtained with anodic spark deposition. Electrochemically treated titanium showed promising results and was able to introduce substantial improvements in achieving fast and stable osseointegration of implants in osteopenic

 $_{228}$ sheep bone 35.

²²⁹ 12 d) Biologic Coatings

Puleo and Nanci (1999) emphasized the importance of biochemical methods of surface modification as an alternative or adjunct to morphologic approaches. Biochemical methods are aimed at control of the tissueimplant interface by the immobilization and/or delivery of proteins, enzymes, or peptides for the purpose of inducing specific cell and tissue responses. They rely on current understanding of the biology and biochemistry of cellular function and differentiation and on suitable surface modification techniques 36 . e) Bio-molecules i.

Laminins 37 Laminins are major proteins in the basal lamina, a protein network foundation for most cells and organs. They are an important and biologically active part of the basal lamina, influencing cell differentiation, migration, adhesion as well as phenotype and survival.

238 ii.

Fibronectin 37 Fibronectin is a high-molecular weight (~440 kDa) extracellular matrix glycoprotein that binds to membrane-spanning receptor proteins called integrins.

In addition to integrins, fibronectin also binds extracellular matrix components such as collagen, fibrin and heparan sulfate proteoglycans.

It is involved in cell adhesion, growth, migration and differentiation. Cellular fibronectin is assembled into the extracellular matrix, an insoluble network that separates and supports the organs and tissues of an organism.

245 iii.

246 13 Vitronectin

Vitronectin is an abundant glycoprotein found in serum the extracellular matrix and promotes cell adhesion and
 spreading.

Vitronectin serves to regulate proteolysis initiated by plasminogen activation. Additionally Vitronectin is a component of platelets and is thus involved in hemostasis. Vitronectin contains an RGD sequence which is a

binding site for membrane bound integrins, e.g. the Vitronectin receptor, which serve to anchor cells to the extra cellular matrix. iv.

²⁵³ 14 RhBMP-2

BMP's are Bone morphogenetic proteins. They are members of -growth and differentiation protein family. They are homodimeric, glycosylated proteins that are highly conserved across species. They are found to be osteoinductive in animals and humans.

They are supposed to promote bone induction by increasing Chemotaxis and increasing the proliferation and differentiation of bone forming cells from undifferentiated mesenchymal cells.

They induce the formation of both trabecular and woven bone. The formed bone remodels based on the demand at the particular site. The delivery of BMPs is aimed at local administration, which is in favor for coating the implant surfaces. v.

²⁶² 15 Bio molecules and Implants

The proportions of these biologic molecules and the presence of other lesser-known components seem to vary with the anatomic location and specific function of the individual basement membrane. Ultra structural data provided by Swope and James (1981) indicate that hemidesmosomes formed on Vitallium Volume XIV Issue IV Version I Year () J implants in monkeys after 2 days and became well established after 3 days 38.

However, more recently published data dispute these findings, indicating that hemidesmosomal contacts were found only on apatite and polystyrene substrates.

269 vi.

270 16 Amino acid sequence RGD

In a goat femur wound chamber model, Bernhardt et al. (??005) compared bone-to-implant contact on uncoated

titanium implant surfaces with RGD peptide-coated surfaces. After 5 and 12 weeks of healing, no significant effect of RGD coating on the mean bone-to-implant contact percentages was observed 39. These results contradict the findings of

$_{275}$ 17 Schliephake et al. (2005b

276) compared, in the mandible of dogs, machined titanium implant surfaces (Ti) with RGD-coated implant surfaces.
277 RGD coatings were achieved either with low RGD concentrations (100 m mol/ml) (RGD low') or with high RGD
278 concentrations (1000 m mol/ml) (RGD high). After 1 month of healing, bone-to-implant contact was significantly
279 high ror RGD high compared with Ti. After 3 months of healing, boneto-implant contact was significantly higher
280 for RGD high and for RGD low compared with Ti 40.

281 vii.

²⁸² 18 Collagen and collagen mimetic peptides

The in vivo osteoconductive potential of type I collagen, type III collagen and collagen mimetic peptide sequences as coating for titanium implants was investigated in the publications of Rammelt et al. (2004,2006,2007) [39][40][41][42].

In the proximal tibial metaphysis of rats, Reyes et al. (2007) compared the mechanical anchorage as well as bone-to-implant contact of machined c.p.titanium implant surfaces (Ti)with either bovine type I collagen (Col-I) or glycine-phenyl alanine-hydroxy proline-glycine-glutamate-arginine (GFOGER; a collagen mimetic peptide sequence)-coated implant surfaces. After 4 weeks of healing, the mean pull-out forces were around 35N for GFOGER, 20N for ColI and 35N for Ti. GFOGER was statistically higher compared with ColI or Ti, but the values for Col I were not statistically higher compared with Ti. The authors concluded that both coatings (GFOGER and ColI) enhanced bone repair and implant integration. viii.

²⁹³ 19 Collagen composite coating with CaP

In the mandible of dogs, Schliephake et al. (??003) compared bone-to-implant contact between titanium alloy implants with a polished surface (Ti), collagen-coated (Col), mineralized (hydroxyapatite) collagen-coated (Col/HA), sequentially hydroxyapatitecollagen-coated (Col/sew HA) and hydroxyapatitecoated titanium surfaces (HA). Animals were sacrificed after 1 and 3 months of healing. No significant differences in the mean bone-toimplant contact between the various implant surfaces were observed in cortical as well as in cancellous bone after

²⁹⁹ 1 and 3 months of implantation 43. ix.

300 20 Growth factor coatings

Growth factors are signaling proteins that promote replication, differentiation, protein synthesis and /or migration of appropriate cell types. In case of endosseous titanium implants, an enhanced proliferation and differentiation of undifferentiated mesenchymal cells osteoprogenitor cells and preosteoblasts into osteoblasts may enhance bone

³⁰⁴ healing (Chappard et al.1999) 44.

Therefore, the rational to coat titanium implants with locally acting growth factors is the assumption that the release of these growth factors might improve the remodeling process at the bone-implant interface, leading to enhanced bone response (De Jonge et al. 2008) **??**5. x.

308 21 Bone Morphogenic Proteins

A particular class of growth factors, BMPs, has shown considerable potential to stimulate bone formation both in extra skeletal sites (Yamazaki et al. 1996; Yoshida et al. 1998) While BMP-2 is used more commonly, BMP-4 is also considered as a candidate growth factor that might improve the remodeling process at the bone-implant interface (Stadlinger et xii.

313 22 Bone-like coatings

A method to self-assemble and mineralize collagen gel and to precoat a bone-like layer of J mineralized collagen immobilized on titanium implant surfaces has been demonstrated. The mineralized layer was found to promote cellular activity, indicating potential for more efficient bone remodeling at the implanttissue interface. This may promote and/or accelerate osseointegration 58.

318 **23** VII.

- 319 Removal of Material (Subtractive Methods)
- Implant Surfaces can be roughened by various material removing techniques. Of which the most common methods are:
- 322 1. Sandblasting. 2. Acid etching.

323 24 Machining. a) Machining

324 The machining of Cp titanium imparts a surface roughness that is distinct from smooth or polished surfaces.

The machining method is an important determinant of the resulting surface. Different surfaces are imparted by machining or subsequent modification. Electro polishing of machined components can further reduce variations

327 measured at the surface, but such surfaces are not well osseointegrated. Creating topographic variation from

the mean surface plane can be achieved by abrasion (TiO2 blasting or soluble/resorbable blasting materials [S/RBM]), blasting, blasting and etching (alumina oxide and H2SO4/HCl), anodizing, cold working (dimpling), and different chemical etching methods (H2SO4/HCl)59-61. Bone to implant contact is one of the important factors for osseointegration. Bone to implant contact is higher for osteotite surfaces when compared to machined surfaces 62.

333 25 b) Grit Blasting

Another approach for roughening the titanium surface consists in blasting the implants with hard ceramic 334 particles. The ceramic particles are projected through a nozzle at high velocity by means of compressed air. 335 Depending on the size of the ceramic particles, different surface roughnesses can be produced on titanium implants. 336 The blasting material should be chemically stable, biocompatible and should not hamper the osseointegration of 337 the titanium implants. Various ceramic particles have been used, such as alumina, titanium oxide and calcium 338 phosphate particles. Alumina (Al2O3) is frequently used as a blasting material and produces surface roughness 339 varying with the granulometry of the blasting media. However, the blasting material is often embedded into the 340 implant surface and residue remains even after ultrasonic cleaning, acid passivation and sterilization. Alumina is 341 insoluble in acid and is thus hard to remove from the titanium surface. In some cases, these particles have been 342 released into the surrounding tissues and have interfered with the osseointegration of the implants. Moreover, 343 this chemical heterogeneity of the implant surface may decrease the excellent corrosion resistance of titanium in 344 a physiological environment 63. 345

³⁴⁶ 26 c) Acid-etching

Etching with strong acids such as HCl, H2SO4, HNO3 and HF is another method for roughening titanium dental
 implants.

Acid etching produces micro pits on titanium surfaces with sizes ranging from 0.5 to 2nm in diameter. 64 Acid-Immersion of titanium implants for several minutes in a mixture of concentrated HCl and H2SO4 heated above100 ?C (dual acid-etching) is employed to produce a micro rough surface. This type of surface promotes rapid osseointegration while maintaining long-term success over 3 years ??5. Enhanced bony anchorage was noted to dual acidetched implants as compared to machined implants 66.

Acid-etched implants showed significantly higher mineral apposition rates compared to acidetched, phosphate 354 coated implants 67. d) Sand Blasted and Acid etched (SLA) surface Among the various techniques to produce 355 a micro rough titanium surface, the combination of sand blasting and acid etching can be used. These surfaces 356 showed enhanced bone apposition in histomorphometric studies, and higher torque values in biomechanical 357 testing. Based on these experimental studies, clinical studies were initiated to load SLA implants after a reduced 358 healing period of only 6 weeks. The clinical examination up to 3 years demonstrated favorable results, with success 359 rates around 99% ??8. e) Chemically Modified SLA Surface: SLActive SLActive is based on the scientifically 360 proven SLAR topography (M. de Wild 2004.). In addition, it has a fundamentally improved surface chemistry. 361 The chemically active, hydrophilic SLActive surface promotes the initial healing reaction, allowing for direct cell 362 interaction at the initial stage of the osseointegration process. Bone formation is immediately initiated resulting 363 in earlier secondary stability and reducing the critical dip. D.Buser et al studied the modified SLA surface 364 produced by rinsing under N2 protection and storing in an isotonic NaCl solution. They demonstrated that the 365 modSLA surface promoted enhanced bone apposition during early stages of bone regeneration 69. 366

Michael M. Bornstein et al showed that Dental implants with a mod SLA surface (SLActive) demonstrated 367 statistically significant differences for probing depths and clinical attachment level values compared to the 368 historic control group, with the mod SLA surface implants having overall lower probing depths and clinical 369 attachment level scores 70. Figure -9. Modification of Material a) Ion Implantation CO ion implantation is 370 371 a new surface treatment designed to improve implant bone integration by modifying the chemical structure of the implant surface at the atomic level without adding or removing material. This is a high vacuum physical 372 technique (<10-4Pa) in which the surface of a material is bombarded with previously selected and accelerated 373 ions that become integrated or implanted within the outer atomic layers of the surface, thereby modifying the 374 physicochemical properties. A study showed improved percentage BIC values for implants with ion-implanted 375 surfaces in comparison to Diamond like Carbon coating and machined controls. Furthermore, bone integration 376 appeared to be accelerated in the ion implantation group 71. 377

378 27 b) Optimum Roughness

The topography of rough surfaces is characterized by different surface roughness parameters (Ra, Rq, Rt, Rsk, Rku, ?q, or ?q, in 3D or 3D mode). Hansson described that an average surface roughness Ra (filtering 50x50 µm) of about 1.5 µm gave the strongest fixation for a bone-metal interface. If the implants are smoother or rougher than this, the anchorage between bone and implant decreases.

A typical measure of implant surface roughness is the Ra value: the arithmetic mean value of the surface departures from the mean plane. Unfortunately, surfaces may have very different morphologies and still share a common Ra value. It is clear that height descriptors alone do not adequately describe surface roughness. More recently, the average peak spacing (Sa) has been associated with implant behavior. There is enough evidence

31 RECENT INNOVATIONS AND FUTURE DIRECTIONS A) NANOTECHNOLOGY

for the positive relationship between surface roughness and increased bone to implant contact (BIC) **??2**. c) Optimal Surface Properties Surface properties of implants directly influence bone responses. Thus, irrespective of the surface modification technology used, detailed surface characterization of an implant is important. Based on the bone response in the present study, which was expressed as a function of quantitative changes in the surface oxide properties, the following appear to be the optimum surface properties of oxidized implants:

? The optimal oxide thickness of a porous surface structure appeared to be in the range of 1,000 to 5,000 nm.
An optimum porosity of open pores is in the range of 19% to 30%, (i.e.) approximately 24%; with a pore size of 2.0 ?m.
Surface roughness values of 0.7 to 1.0 ?m for Sa, 0.9 to 1.4 ?m for Sq, and 27% to 46% for Sdr seemed to be optimum.
TiO2 in a crystalline phase seemed to be optimal 73.

³⁹⁶ 28 d) Controversies

With Respect To Implant Topographies Machined titanium surfaces have been reported to favor fibroblastic 397 growth, migration, and spread, and therefore were considered favorable for formation of peri-implant soft tissue. 398 On the other hand, because of the increased proliferative activity of fibroblasts on machined surfaces, fibrous 399 400 capsules or connective tissue overgrowth can form, compromising local blood supply and leading to failure of the 401 implant to integrate with the soft tissue. To overcome this problem, rough Rough titanium surfaces have been 402 reported to improve attachment and decrease growth and spread of fibroblasts. However, a diminished growth of fibroblasts on rough titanium surfaces can result in the formation of a thin connective tissue that will not be 403 404 capable of supporting surrounding tissue structures. In addition, rough implant surfaces have been reported to 405 be especially prone to peri-implant infection and seem also to attract inflammatory cells.

Another suggested titanium surface comprises grooved topography, which has been demonstrated to favor the orientation and alignment of fibroblasts and claimed in several studies to be appropriate for the establishment of an organized connective tissue structure around the implant. However, the exact topographic configurations of grooved titanium surfaces that are appropriate for the in vivo establishment of longterm stable and overall optimal peri-implant soft tissue conditions are still largely unknown.

There is a lack of knowledge about the ideal implant surface characteristics that lead to the establishment of optimal connective tissue and attachment around titanium implants. The acid-etching and blasting methods generally do not change the main compositional surface elements of the titanium, which consist mainly of titanium and oxygen, but rather the surface morphology/topography and consequently surface roughness, two action mechanisms of osseointegration of oxidized implants have been proposed: 1) mechanical interlocking through bone growth in pores and 2) biochemical bonding [74][75].

⁴¹⁷ 29 e) Surface roughness at the nano scale level

The chemistry and roughness of implant surfaces play a major role in the biological events that follow implantation. Nevertheless, surfaces are often developed using an empirical approach with in vitro and in vivo tests. Most of the surfaces currently available have random topography with a wide range of thicknesses, from nanometers to millimeters 76.

The exact biological role of these features is unknown because of the absence of standardized surfaces with repetitive topography at the nano-sized level (e.g. pits with fixed diameters and depth, lanes with controlled profiles). Such controlled or standardized surfaces might help to understand the interactions between specific proteins and cells. These standardized surfaces might also promote early bone apposition on the implants.

426 Only a few studies have reported modifications to the roughness as well as the chemistry at the nanometer scale 427 in a reproducible manner. Most of these attempts have used processing methods from the electronic industry 428 such lithography and surface laserpitting.

These nanometer structures may also give the cells positive guidance by means of the selective attachment of osteoblasts to the implant surface. This selective attachment process might result in the improvement of initial healing around dental implants22. f) Re-Osseo integration Persson et al (2001) evaluated reosseointegration of SLA (Sandblasted and acid etched) and turned implants in dogs. They found that reosseointegration was substantial for implants with SLA surfaces but only minimal for exposed smooth (turned) surfaces. Reosseointegration (BIC) at SLA surfaces averaged 84% compared to 22% at turned implant surfaces 77.

435 **30 IX.**

⁴³⁶ 31 Recent Innovations and Future Directions a) Nanotechnol-

437 **Ogy**

Asson Nanotechnology is the engineering of functional systems at the molecular scale. Materials reduced to the nanoscale can show different properties compared to what they exhibit on a macro scale, enabling unique applications. For instance, opaque substances become transparent (copper); stable materials turn combustible (aluminum); insoluble materials become soluble (gold). A material such as gold, which is chemically inert at normal scales, can serve as a potent chemical catalyst at nanoscale. Much of the fascination with nanotechnology stems from these superturn and surface phonemene that matter archibits of the papersel.

these quantum and surface phenomena that matter exhibits at the nanoscale.

Nanotechnology involves materials that have a nano-sized topography or are composed of nano-sized materials. These materials have a size range between 1 and 100 nm (109m) Nanotechnology often involves onedimensional concepts (nano-dots and nano wires) or the self-assembly of more complex structures (nanotubes). Materials are also classified according to their form and structure as nanostructures, nanocrystals, nano coatings. b) Methods of Creating Nano-topography Nanotechnology requires novel ways of manipulating matter in the atomic scale. Several approaches are currently prevalent in the experimental application to endosseous implants.

1. One approach involves the physical method of compaction of nano-particles of TiO2 vs. micronlevel 450 particles to yield surfaces with nano scale grain boundaries78. An advantage of this method is that it conserves 451 the chemistry of the surface among different topographies. 2. Second is the process of molecular selfassembly. Self-452 assembled monolayers (SAMs) are formed by the spontaneous chemisorptions and vertical close-packed positioning 453 of exposed functional end group could be an osteo inductive or cell adhesive molecule. An example of this is the 454 use of cell adhesive peptide domains (RGD domains) appended to SAMs composed of poly ethylene glycol (PEG) 455 456 and applied to the titanium implant surfaces. 3. A third method is the chemical treatment of different surfaces to expose reactive groups on the material surface and create nano-scale topography. This is popular among current 457 dental implant investigators. NaOH treatment catalyzes the production of titanium nanostructures outward from 458 the titanium surface 79. 4. The deposition of nanoparticles on to the titanium surface represents a fourth approach 459 to imparting nanofeatures to a titanium dental implant80 Solgel transformation techniques achieve deposition 460 of nano meter-scale calcium phosphate accretions to the implant surface 81-82. Alumina, Titania, zirconia and 461 462 other materials can also be applied 83. Owing to their resultant atomic-scale interactions, the accretions display 463 strong physical interactions. a. In a modified approach, Nishimura and colleagues [2007] demonstrated a directed 464 approach to assembly of CaPO4 nano features on dual acid-etched cp Titanium implant surfaces. The deposition 465 of discrete 20-40nm nanoparticles on an acid-etched titanium surface led to increased mechanical interlocking with bone and the early healing of bone at the endosseous implant surface in a rat model. One of the main 466 concerns related to coating the implant surface is the risk of coating detachment and toxicity of related debris 467 84. 5. A fifth approach to creating nano scale topography on Titanium is the use of optical methods (typically 468 lithography) reliant on wave length specific dimensions to achieve the appropriate nano scale modification. These 469 470 approaches are labor intensive methods that require considerable development prior to clinical translation. The present use of lasers to promote micron-level groove on an implant surface can produce micron-level, not nano 471 scale, modification of the implant surface. Another method of depositing nano scale material on to the implant 472 surface involves ion beam deposition (e.g. hydroxyapatite) 85. 473 Х. 474

475 **32** Conclusion

Implant surface characteristics are widely recognized as being of fundamental importance in achieving long-476 term implant success. As such, extensive research has been performed in order to determine the surface 477 texture necessary to attain an optimal bone-implant biomechanical interlock. Four interrelated properties of an 478 implant surface affect osteogenic activity: chemical composition, surface energy, surface roughness, and surface 479 morphology.Osseointegration and its underlying mechanisms of cell attachment, migration, proliferation, and 480 differentiation are sensitive to one or more of these properties. Methods of enhancing the implant surface include 481 alteration of the microstructure and modification of its physiochemical parameters, including surface free energy 482 and wettability. 483

The surface qualities are of utmost importance in establishing of a reaction between the implant and the tissues. This concerns the surface structure as well as its chemical and biological properties. Much attention has been focused on the importance of the macrostructure of the implants for establishing retention in the bone. More attention will probably be focused in the future on the biological effects of the surface structure on the microstructural and ultrastructural levels as well as on the surface chemistry of the implants. Progress in these fields based on knowledge of the biological effects may provide implants with improved tissue response and clinical performance in the future.



Figure 1:



23

Figure 2: Figure 2 : Figure 3 :



Figure 3: Figure 4 :



Figure 4:



Figure 5: Figure 5 :



Figure 6:



Figure 7: Figure 6 :



Figure 8:



Figure 9:



Figure 10: Figure 7 :





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32 CONCLUSION

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